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Schrödinger equation in terms of linearly averaged position moments

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Linearly averaged position moment (LAPM) $\begin{bmatrix} R_1^{k_1}R_2^{k_2}\cdots R_{3N}^{k_{3N}} \end{bmatrix}$ is defined as the linear (not quadratic) average of the position moment operator $R_1^{k_1}R_2^{k_2}\cdots R_{3N}^{k_{3N}}$ over the *N*-electron wave function $\Psi(\{R_j\})$, where $\{R_j\}$ are 3*N* Cartesian coordinates of electrons and $\{k_j\}$ are non-negative integers. When all the LAPM's are well defined, it is shown that the Schrödinger equation is equivalent to a set of an infinite number of equations between LAPM's involving the potential-energy operator. The kinetic energy operator does not appear. The spherical polar representation of the LAPM equation is also presented. Illustrations are given for simple one- and two-electron atoms, where the LAPM equation is applied to the determination of approximate wave functions and associated energies.

I. INTRODUCTION

Previously, the zero momentum energy expression has been proposed¹⁻³ and applied¹⁻⁵ as a sensitive criterion to assess the accuracy of approximate wave functions. The expression has been derived by the application of the local energy concept^{6,7} to the momentum-space Schrödinger equation: The equation to be satisfied at the momentumspace origin is considered and is inversely transformed into position space. The resultant zero momentum energy expression includes rather strange integrals,

$$\left[\prod_{j=1}^{3N} R_{j}^{k_{j}}\right] \equiv \int \left(\prod_{j=1}^{3N} R_{j}^{k_{j}}\right) \Psi(\mathbf{r}) d\mathbf{r}, \qquad (1a)$$

$$\left[\left(\prod_{j=1}^{3N} R_{j}^{k_{j}}\right) V(\mathbf{r})\right] \equiv \int \left(\prod_{j=1}^{3N} R_{j}^{k_{j}}\right) V(\mathbf{r}) \Psi(\mathbf{r}) d\mathbf{r}, \quad (1b)$$

which are linear averages of the Cartesian moment $\prod_{j=1}^{3N} R_j^{k_j}$ and potential-energy $V(\mathbf{r})$ operators over the position wave function $\Psi(\mathbf{r})$ (see also Ref. 8). $\{k_j\}$ are non-negative integers.

Stimulated by these studies, we have recently shown⁹ that there exists a set of relations between position moments linearly averaged over the wave function. These relations constitute a necessary condition for the true wave function, and the zero momentum energy formula¹⁻³ follows from the simplest case of the relations. A further generalization is proposed¹⁰ for the linear average over the wave function in a finite domain, whose results can be used as a measure for the regional accuracy of wave functions. These studies have added nontrivial information to the structure of the Schrödinger equation.

In the present paper, we show that the set of relations between linearly averaged position moments (LAPM's) has a *deterministic* property. When all the LAPM's are well defined, the Schrödinger equation is equivalent to a set of an infinite number of equations between LAPM's involving the potential-energy operator $V(\mathbf{r})$. The kinetic energy operator becomes implicit. In the next section, the formalism is presented. Some illustrative applications are given in Sec. III for the determination of approximate wave functions and associated energies from the LAPM equation. Hartree atomic units are used throughout this paper.

II. FORMALISM

The N-electron Schrödinger equations in position and momentum spaces are written as

$$(-\Delta/2 + V(\mathbf{r}) - E)\Psi(\mathbf{r}) = 0, \qquad (2a)$$

$$(p^2/2 - E)\Phi(\mathbf{p}) + \int W(\mathbf{p} - \mathbf{p}')\Phi(\mathbf{p}')d\mathbf{p}' = 0, \quad (2b)$$

where $\mathbf{r} \equiv \{\mathbf{r}_j\}$ and $\mathbf{p} \equiv \{\mathbf{p}_j\}$ are position and momentum vectors of the electrons, respectively, and $\Delta \equiv \sum_{j=1}^{N} \Delta_j$ and $p^2 \equiv \sum_{j=1}^{N} |\mathbf{p}_j|^2$. The wave functions $\Psi(\mathbf{r})$ and $\Phi(\mathbf{p})$ and the potential-energy operators $V(\mathbf{r})$ and $W(\mathbf{p})$ are, respectively, pairs of the Fourier transforms:

$$\Phi(\mathbf{p}) = (2\pi)^{-3N/2} \int \exp(-i\mathbf{r}\cdot\mathbf{p})\Psi(\mathbf{r}) d\mathbf{r}, \qquad (3a)$$

$$\Psi(\mathbf{r}) = (2\pi)^{-3N/2} \int \exp(+i\mathbf{r}\cdot\mathbf{p})\Phi(\mathbf{p}) d\mathbf{p}, \qquad (3b)$$

$$W(\mathbf{p}) = (2\pi)^{-3N} \int \exp(-i\mathbf{r}\cdot\mathbf{p}) V(\mathbf{r}) d\mathbf{r}, \qquad (4a)$$

$$V(\mathbf{r}) = \int \exp(+i\mathbf{r}\cdot\mathbf{p}) W(\mathbf{p}) d\mathbf{p}, \qquad (4b)$$

where **r**•**p** means $\sum_{i=1}^{N} \mathbf{r}_i \cdot \mathbf{p}_i$.

Combining Eqs. (2)-(4), we obtain the Schrödinger equation in an intermediate representation:

$$(p^2/2 - E)[\exp(-i\mathbf{r}\cdot\mathbf{p})] + [\exp(-i\mathbf{r}\cdot\mathbf{p})V(\mathbf{r})] = 0, (5)$$

where the square brackets stand for

$$[f(\mathbf{r})] \equiv \int f(\mathbf{r}) \Psi(\mathbf{r}) \, d\mathbf{r},\tag{6}$$

which is a *linear* average of $f(\mathbf{r})$ over the position wave function $\Psi(\mathbf{r})$. We implicitly assume that the two integrals appearing in Eq. (5) do exist.

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A. Cartesian representation

Now we expand the exponential function $\exp(-i\mathbf{r}\cdot\mathbf{p})$ in an absolutely convergent series:

$$\exp(-i\mathbf{r}\cdot\mathbf{p}) = \prod_{j=1}^{3N} \sum_{k_j \neq 0}^{\infty} (-iR_j P_j)^{k_j} / k_j!$$

=
$$\sum_{\{k_j\}=0}^{\infty} (-i)^{\sum_j k_j} (R_1 P_1)^{k_1} \cdots (R_{3N} P_{3N})^{k_{3N}} / k_1! \cdots k_{3N}!,$$
 (7)

where $\{R_j\}$ and $\{P_j\}$ (j = 1, 2, ..., 3N) represent the 3N Cartesian coordinates of the position and momentum vectors of the N electrons. We assume all the LAPM's $[\prod_{j=1}^{3N} R_j^{k_j}]$ and $[(\prod_{j=1}^{3N} R_j^{k_j}) V(\mathbf{r})]$ exist and their linear combinations are well defined. Then we can exchange the integration and the summation when Eq. (7) is substituted into Eq. (5), and we have

$$\sum_{\{k_{j}\}=0}^{\infty} (-i)^{\sum_{j}k_{j}} \left\{ \frac{1}{2} \sum_{m=1}^{3N} k_{m} (k_{m}-1) \left[R_{1}^{k_{1}} \cdots R_{m-1}^{k_{m-1}} R_{m}^{k_{m}-2} R_{m+1}^{k_{m+1}} \cdots R_{3N}^{k_{1N}} \right] + E \left[R_{1}^{k_{1}} \cdots R_{3N}^{k_{1N}} \right] - \left[(R_{1}^{k_{1}} \cdots R_{3N}^{k_{3N}}) V(\mathbf{r}) \right] \right\}$$

$$\times (P_{1}^{k_{1}} \cdots P_{3N}^{k_{3N}}) / (k_{1}! \cdots k_{3N}!) = 0, \qquad (8)$$

as an alternative form of Eq. (5). For Eq. (8) to hold for any values of $\{P_j\}$, all the coefficients of $P_1^{k_1} \cdots P_{3N}^{k_{3N}}$ appearing on the left-hand side of Eq. (8) must vanish;

$$\frac{1}{2}\sum_{m=1}^{3N} k_m (k_m - 1) \left[R_m^{-2} \prod_{j=1}^{3N} R_j^{k_j} \right] + E \left[\prod_{j=1}^{3N} R_j^{k_j} \right] - \left[\left(\prod_{j=1}^{3N} R_j^{k_j} \right) V(\mathbf{r}) \right] = 0,$$
(9)

where $k_j = 0, 1, 2, ...$. Equation (9) represents a set of an infinite number of equations, and is equivalent to the Schrödinger equation provided that all the LAPM's and their linear combinations are well defined. It may be interesting to observe that in Eq. (9) the kinetic energy operator does not appear and its contribution is represented by the first term.

If we set
$$k_2 = k_3 = \cdots = k_{3N} = 0$$
 in Eq. (9), we obtain a special form:
 $\frac{1}{2}k_1(k_1 - 1)[R_1^{k_1 - 2}] + E[R_1^{k_1}] - [R_1^{k_1}V(\mathbf{r})] = 0,$
(10)

which is a set of single-component equations and constitutes a subset of Eq. (9).

B. Spherical polar representation

When the system under consideration has a spherical symmetry, the use of the spherical polar coordinates is natural. The relation to be satisfied between the LAPM's in the spherical polar coordinates can be derived as follows.

We first invoke the plane-wave expansion¹¹

$$\exp(-i\mathbf{p}_{j}\cdot\mathbf{r}_{j}) = 4\pi \sum_{l_{j}=0}^{m} \sum_{m_{i}=-l_{j}}^{l_{j}} (-i)^{l_{j}} j_{l_{j}}(p_{j}r_{j}) Y_{l_{j}m_{j}}(\Omega_{j}) Y_{l_{j}m_{j}}^{*}(\omega_{j}),$$
(11)

where (p_j, Ω_j) and (r_j, ω_j) are the spherical polar coordinates of the vectors \mathbf{p}_j and \mathbf{r}_j , respectively. Y_{lm} is the usual spherical harmonic¹¹ and $j_l(z)$ is the spherical Bessel function of the first kind.¹² Substituting Eq. (11) into the parent equation (5) and gathering the coefficients of the linearly independent term $\prod_{j=1}^{N} Y_{l,m_j}(\Omega_j)$, we obtain

$$\frac{1}{2} \left(\sum_{i=1}^{N} p_{i}^{2} \right) \left[\prod_{j=1}^{N} j_{l_{j}}(p_{j}r_{j}) Y_{l_{j}m_{j}}^{*}(\omega_{j}) \right] - E \left[\prod_{j=1}^{N} j_{l_{j}}(p_{j}r_{j}) Y_{l_{j}m_{j}}^{*}(\omega_{j}) \right] + \left[V(\mathbf{r}) \prod_{j=1}^{N} j_{l_{j}}(p_{j}r_{j}) Y_{l_{j}m_{j}}^{*}(\omega_{j}) \right] = 0.$$
(12)

Substitution of the expansion¹² for $j_1(z)$

$$j_{l}(z) = (2z)^{l} \sum_{k=0}^{\infty} \left\{ (-1)^{k} (l+k)! / k! (2l+2k+1)! \right\} z^{2k},$$
(13)

into Eq. (12) followed by comparison of the coefficients of $\prod_{j=1}^{N} p_{j}^{l_{j}+2k_{j}}$ gives the desired result:

$$\sum_{i=1}^{N} k_i (2l_i + 2k_i + 1) \left[r_i^{-2} \prod_{j=1}^{N} r_j^{l_j + 2k_j} Y_{l_j m_j}^*(\omega_j) \right] + E \left[\prod_{j=1}^{N} r_j^{l_j + 2k_j} Y_{l_j m_j}^*(\omega_j) \right] - \left[V(\mathbf{r}) \prod_{j=1}^{N} r_j^{l_j + 2k_j} Y_{l_j m_j}^*(\omega_j) \right] = 0,$$
(14)

where $k_i = 0, 1, 2, ...; l_i = 0, 1, 2, ...; m_i = -l_i, -l_i + 1, ..., + l_i$ for i = 1, 2, ..., N. Equation (14) is a set of LAPM equations in the spherical polar coordinates.

A special case of Eq. (14) for $l_i = m_i = 0$ (i = 1,2,...,N) reads

$$\sum_{i=1}^{N} k_i (2k_i + 1) \left[r_i^{-2} \prod_{j=1}^{N} r_j^{2k_j} \right] + E \left[\prod_{j=1}^{N} r_j^{2k_j} \right] - \left[\left(\prod_{j=1}^{N} r_j^{2k_j} \right) V(\mathbf{r}) \right] = 0,$$
(15)

since $Y_{00}(\omega_j) = (4\pi)^{-1/2}$. If we set $k_2 = k_3 = \cdots = k_N = 0$, Eq. (15) further reduces to a set of single-component equations

$$k_{1}(2k_{1}+1)\left[r_{1}^{2k_{1}-2}\right]+E\left[r_{1}^{2k_{1}}\right]-\left[r_{1}^{2k_{1}}V(\mathbf{r})\right]=0,$$
(16)

which may compare with Eq. (10) in the Cartesian representation.

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Equation (9) may be considered as a projection of the position-space Schrödinger equation onto momentum space with the discrete Cartesian bases $\prod_{j=1}^{3N} P_j^{k_j}$ defined by a set of non-negative integers $\{k_j\}$. Similarly, Eq. (14) is regarded as a projection onto the spherical polar bases $\prod_{j=1}^{N} p_j^{l_j+2k_j} Y_{l_jm_j}(\Omega_j)$ defined by the integers $\{k_j, l_j, m_j\}$. We can then use Eq. (9) or (14) for the determination of an approximate wave function and the associated energy. When a trial function includes M parameters, M + 1 equations from the set (9) or (14) [or the subset (10), (15), or (16)] determine the values of these parameters as well as the approximate energy E. In this approach, no existing approximation theories are invoked such as the variation method and the perturbation theory.

Then a problem of how to choose the M + 1 equations from the set (9) or (14) emerges. In the case of the Cartesian representation, we should be reminded that an equation from Eq. (9) with some $\{k_j\}$ constitutes the coefficient of the bases $\prod_{j=1}^{N} P_j^{k_j}$ when the Schrödinger equation (5) is expanded. If we adopt equations with smaller (larger) $\{k_j\}$, the Schrödinger equation is satisfied in a smaller (larger) **p** region of momentum space. Therefore the choice depends on which region one needs more accurate description for. Here we note that the position and momentum representations emphasize inverse regions of the respective spaces. Analogous discussion holds for the set (14) given in the spherical polar coordinates.

III. ILLUSTRATIVE APPLICATIONS

In the following, we apply the LAPM equation to (approximate) solutions of two simple systems, i.e., the hydrogen-like atom and the ground-state helium atom. In the former case, the correct wave function and the associated eigenenergy are completely reproduced. In the latter case, we discuss the property of approximate LAPM solutions using a simple trial function.

A. Hydrogen-like atom

For a single-electron system, the LAPM equation in the spherical polar coordinates [Eq. (14)] reads

$$k(2k+2L+1)[r^{2k+L-2}Y^*_{LM}(\omega)] + E[r^{2k+L}Y^*_{LM}(\omega)] - [V(\mathbf{r})r^{2k+L}Y^*_{LM}(\omega)] = 0.$$
(17)

When $V(\mathbf{r}) = -Z/r$ (i.e., a hydrogen-like atom with nuclear charge Z), Eq. (17) suggests that the wave function $\Psi(\mathbf{r})$ has the following form:

$$\Psi(\mathbf{r}) = R(r)Y_{im}(\omega). \tag{18}$$

Substituting Eq. (18) and the explicit form of $V(\mathbf{r})$ into Eq. (17), we obtain the radial LAPM equation:

$$k(2k+2l+1)[r^{2k+l-2}]_r + E[r^{2k+l}]_r + Z[r^{2k+l-1}]_r = 0,$$
(19)

where

$$[f(r)]_{r} \equiv \int_{0}^{\infty} f(r) R(r) r^{2} dr.$$
(20)

If we assume a radial function R(r) of the form

$$R(r) = \left(\sum_{i=0}^{n-1} a_i \zeta^{i} r^{i}\right) \exp(-\zeta r), \qquad (21)$$

Eq. (19) becomes

$$k(2k+2l+1)\sum_{i=0}^{n-1}a_i(2k+l+i)! + (E/\zeta^2)\sum_{i=0}^{n-1}a_i(2k+l+i+2)! + (Z/\zeta)\sum_{i=0}^{n-1}a_i(2k+l+i+1)! = 0, \quad (22)$$

from which we have to determine E, $\{a_i\}$, and ζ for a given set of Z, l, and n.

To solve Eq. (22), we introduce a linearly independent basis set defined by

$$(x)_{i} \equiv \begin{cases} (x+1)(x+2)\cdots(x+i) & \text{for } i \ge 1, \\ 1 & \text{for } i = 0. \end{cases}$$
(23)

Then Eq. (22) is rearranged as

$$(\frac{1}{2} + E/\zeta^{2})a_{n-1}(2k+l)_{n+1} + \{(\frac{1}{2} + E/\zeta^{2})a_{n-2} + (Z/\zeta - n)a_{n-1}\}(2k+l)_{n} + \sum_{i=2}^{n-1} \{(\frac{1}{2} + E/\zeta^{2})a_{i-2} + (Z/\zeta - i)a_{i-1} - \frac{1}{2}(l-i)(l+i+1)a_{i}\}(2k+l)_{i} + \{(Z/\zeta - 1)a_{0} - \frac{1}{2}(l-1)(l+2)a_{1}\}(2k+l)_{1} - \frac{1}{2}l(l+1)a_{0}(2k+l)_{0} = 0.$$

$$(24)$$

For Eq. (24) to hold for any non-negative integer k, all the coefficients of $\{(2k + l)_i\}$ must vanish. We immediately have $\zeta = Z/n, \quad E = -\zeta^2/2 = -Z^2/2n^2,$ (25)

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and a recursion relation for the expansion coefficients $\{a_i\}$

$$(l-i)(l+i+1)a_i = 2(n-i)a_{i-1}, \quad i = 1,2,...,n-1,$$

$$l(l+1)a_0 = 0.$$
(26a)
(26b)
(26b)

When n < l, l - i is positive and therefore Eq. (26a) has a general solution 1 . 1 / 1 1)11(... Caller . 1.

$$a_{l} = \{2l(l+1)(n-1)!(l-l-1)!(n-l-1)!(l+l+1)!\} \quad \{a_{0}, \dots, a_{n}\} = \{a_{n}, \dots, a_{$$

$$a_i = 0, \quad i = 0, 1, \dots, n-1,$$

and there is no physically meaningful solution for the case of n < l. The well-known quantum condition n > l or l < n - 1 thus emerges. In this case, Eqs. (26a) and (26b) have a general solution

$$a_{i} = \begin{cases} \{(-2)^{i-l}(n-l-1)!(2l+1)!/(n-i-1)!(i-l)!(i+l+1)!\}a_{i} & \text{for } l < i < n-1, \\ 0 & \text{for } 0 < i < l, \end{cases}$$
(28)

where a_1 is an arbitrary nonzero constant.

Now the radial function R(r) [Eq. (21)] becomes

$$R(r) = \left(\sum_{i=l}^{n-1} a_i \zeta^{i} r^{i}\right) \exp(-\zeta r)$$

= $(\zeta r)^{l} \left(\sum_{i=0}^{n-l-1} a_{i+l} \zeta^{i} r^{i}\right) \exp(-\zeta r),$ (29a)

where

$$a_{l+l} = \{(-2)^{l}(n-l-1)!(2l+1)! \\ \times [(n-l-i-1)!(2l+i+1)!l!]^{-1}\}a_{l},$$
(29b)

and $\zeta = Z/n$. If we choose

$$a_{l} = -\left[(n+l-1)! \right]^{2} / \left[(n-l-1)! (2l+1)! \right]$$

in Eq. (29b), the polynomial involved in Eq. (29a) is nothing but the associated Laguerre polynomial $L_{n+1}^{2l+1}(2\zeta r)$. We finally have

$$R(r) = (\zeta r)^{l} L_{n+l}^{2l+1}(2\zeta r) \exp(-\zeta r).$$
(30)

Equations (18), (25), and (30) constitute the complete solutions for the hydrogen-like atom obtained from the present LAPM equation. They are identical to the known results¹³ which usually follow from the solution of the Schrödinger equation as a second-order differential equation.

B. Ground-state helium atom

The LAPM equation (15) for N = 2 is

$$k_{1}(2k_{1}+1)\left[r_{1}^{2k_{1}-2}r_{2}^{2k_{2}}\right] + k_{2}(2k_{2}+1)\left[r_{1}^{2k_{1}}r_{2}^{2k_{2}-2}\right] + E\left[r_{1}^{2k_{1}}r_{2}^{2k_{2}}\right] - \left[r_{1}^{2k_{1}}r_{2}^{2k_{2}}V(\mathbf{r}_{1},\mathbf{r}_{2})\right] = 0, \quad (31)$$

where

$$V(\mathbf{r}_1, \mathbf{r}_2) = -Z/r_1 - Z/r_2 + 1/r_{12}, \qquad (32)$$

for a helium-like atom with nuclear charge Z. Let us consider the approximate wave function given by a single product of the scaled hydrogenic 1s functions:

$$\Psi(\mathbf{r}_1,\mathbf{r}_2) = \exp(-\zeta r_1)\exp(-\zeta r_2), \qquad (33)$$

which was first examined by Kellner¹⁴ in the variational treatment of the ground-state helium atom.

For a special case of $k_1 = k$ and $k_2 = 0$, the explicit evaluation of the integrals appearing in Eq. (31) for the wave function (33) results in a simplified LAPM equation

$$k\zeta^{2} + 2(k+1)E + \{(k+2)Z - 1 + (k+3)2^{-2k-3}\}\zeta = 0, \quad (34)$$

which includes the quantities E and ζ to be determined. In contrast to the case of hydrogen-like atoms, there is no set of E and ζ values which satisfies Eq. (34) for all possible k, implying that the exact wave function is not expressed in the form of Eq. (33).

However, we can employ some selected equations from Eq. (34) for an approximate determination of E and ζ . Let us examine the first two equations resulting from k = 0 and 1. They are

$$E + (Z - 5/16)\zeta = 0$$
, for $k = 0$, (35a)

$$\zeta^{2} + 4E + (3Z - \frac{7}{8})\zeta = 0$$
, for $k = 1$. (35b)

Though we cannot uniquely determine two unknowns from a single equation, Eq. (35a) holds if

$$\zeta = \zeta_0 = Z - \frac{5}{16}$$
 and $E = E_0 = -\zeta_0^2 = -(Z - \frac{5}{16})^2$,
(36a)

which are $\zeta_0 = 1.6875$ and $E_0 = -2.847656$ when Z = 2. Accidentally, Eq. (36a) is identical to the result of the variational calculation as has been discussed in relation to the zero momentum energy expression.¹ Similarly, Eq. (35b) holds if

$$\zeta = \zeta_1 = Z - \frac{7}{24}$$
 and $E = E_1 = -\zeta_1^2 = -(Z - \frac{7}{24})^2$,
(36b)

which become $\zeta_1 = 1.708\ 333$ and $E_1 = -2.918\ 403$ when Z = 2. This approximate energy E_1 is lower than the exact energy 15 - 2.903 724, since the LAPM energy is not an upper bound to the true energy in general. The unique solutions that satisfy both Eqs. (35a) and (35b) are

$$\zeta = \zeta_{01} = Z - \frac{3}{8}$$
 and
 $E = E_{01} = -(Z - \frac{5}{16})(Z - \frac{3}{8}),$ (36c)
which become $\zeta_{01} = 1.625$ and $E_{01} = -2.742188$ for
 $Z = 2.$

In order to clarify the difference in the above three approximate solutions [Eqs. (36a), (36b), and (36c)], it may be convenient to introduce the quantity

$$\Delta(\mathbf{p}_{1},\mathbf{p}_{2}) \equiv (p_{1}^{2}/2 + p_{2}^{2}/2 - E) [\exp(-i\mathbf{r}_{1}\cdot\mathbf{p}_{1} - i\mathbf{r}_{2}\cdot\mathbf{p}_{2})] + [\exp(-i\mathbf{r}_{1}\cdot\mathbf{p}_{1} - i\mathbf{r}_{2}\cdot\mathbf{p}_{2}) V(\mathbf{r}_{1},\mathbf{r}_{2})].$$
(37)

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(27a)

(27b)

δ

 δ_0

The condition $\Delta(\mathbf{p}_1, \mathbf{p}_2) = 0$ is equivalent to the Schrödinger equation in momentum space, and hence the deviation of $\Delta(\mathbf{p}_1, \mathbf{p}_2)$ from zero characterizes the property of an approximate solution. For a particular case of $\mathbf{p}_2 = \mathbf{0}$, Eq. (37) reduces to

$$\delta(\mathbf{p}_1) \equiv \Delta(\mathbf{p}_1, \mathbf{0})$$

= $(p_1^2/2 - E) [\exp(-i\mathbf{r}_1 \cdot \mathbf{p}_1)]$
+ $[\exp(-i\mathbf{r}_1 \cdot \mathbf{p}_1) V(\mathbf{r}_1, \mathbf{r}_2)].$ (38)

For the approximate wave function given by Eq. (33), the difference function $\delta(\mathbf{p})$ is evaluated to be

$$\begin{aligned} (\mathbf{p}) &= \delta(p) = 32\pi^{2} \{ (p^{2} - 2E)\zeta^{-2}(p^{2} + \zeta^{2})^{-2} \\ &- Z\zeta^{-1}(p^{2} + \zeta^{2})^{-1} [\zeta^{-2} + (p^{2} + \zeta^{2})^{-1}] \\ &+ 2\zeta^{2} [L_{6}(p;\zeta,E) + L_{7}(p;\zeta,E) \\ &+ L_{8}(p;\zeta,E)] \}, \end{aligned}$$
(39)

where the explicit forms of the functions $L_j(p;\zeta,E)$ (j = 6,7,8) are given in Ref. 16.

In the case of the helium atom (Z = 2), Fig. 1 shows the $\delta(p)$ functions for the three sets of approximate parameters, $(\zeta_0, E_0), (\zeta_1, E_1), \text{ and } (\zeta_{01}, E_{01}).$ For the first set obtained from the LAPM equation (34) with k = 0, the deviation of $\delta(p)$ from zero is small in the low-momentum region and especially $\delta(0) = 0$. For the second set resulting from the LAPM equation with k = 1, $|\delta(p)|$ decreases in the highmomentum region, but increases remarkably in the low-momentum region. For the third set, the low-momentum behavior of $\delta(p)$ is much improved, but a large deviation in the high-momentum region is observed. These behaviors may be easily understood if we are reminded of the fact that the satisfaction of the k th LAPM equation corresponds to making the coefficient of p^{2k} zero when $\delta(p)$ is expanded in a power series of p^2 . In fact, the $\delta(p)$ functions for the three sets of parameters can be expressed as

$$\delta_0(p) = 32\pi^2 [(2^{24}/3^{21})p^2 - (37\cdot2^{29}/3^{27})p^4 + O(p^6)], \qquad (40a)$$

$$\delta_1(p) = 32\pi^2 [(163 \cdot 2^{13} \cdot 3^4/41^5)]$$

 $-(79\cdot 2^{20}\cdot 3^8/41^9)p^4 + O(p^6)], \qquad (40b)$

$$P_1(p) = 32\pi^2 [-(21\cdot 2^{20}/13^9)p^4 + O(p^6)].$$
 (40c)

The low-momentum superiority of the third set [Eq. (36c)] is clear from Eq. (40c). If we adopt a more flexible trial function with several adjustable parameters, we may be able to extend the zero deviation of the $\delta(p)$ function to the intermediate-momentum region. Such a possibility will be examined in the future.



FIG. 1. Behavior of the difference function $\delta(p)$ for the three sets of parameters ζ and E with Z = 2. The solid line is for ζ_0 and E_0 [Eq. (36a)], the dashed line for ζ_1 and E_1 [Eq. (36b)], and the dotted line for ζ_{01} and E_{01} [Eq. (36c)].

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- B. H. Armstrong, Bull. Am. Phys. Soc. 9, 401 (1964); Lockheed Missiles and Space Company Technical Report, Physics, No. 6-74-64-26, 1964.
- ²A. J. Thakkar and V. H. Smith, Jr., Phys. Rev. A 18, 841 (1978).
- ³T. Koga, K. Ohta, and A. J. Thakkar, Phys. Rev. A 37, 1411 (1988).
- ⁴R. J. Drachman, Phys. Rev. **136**, A 641 (1964).
- ⁵T. Koga, J. Chem. Phys. 83, 6301 (1985).
- ⁶J. H. Bartlett, Phys. Rev. 51, 661 (1937).
- ⁷A. A. Frost, J. Chem. Phys. 10, 240 (1942); A. A. Frost, R. E. Kellog, and
- E. C. Curtiss, Rev. Mod. Phys. 32, 313 (1960).
- *T. Koga and A. J. Thakkar, Int. J. Quantum Chem. 34, 103 (1988).
- ⁹T. Koga, J. Chem. Phys. 89, 5961 (1988).
- ¹⁰T. Koga, J. Chem. Phys. 90, 5887 (1989).
- ¹¹A. Messiah, *Quantum Mechanics* (North-Holland, Amsterdam, 1961), Vol. 1, pp. 494–497.
- ¹²M. Abramowitz and I. A. Stegun, *Handbook of Mathematical Functions* (Dover, New York, 1972), p. 435 ff.
- ¹⁴See, for example, L. Pauling and E. B. Wilson, *Introduction to Quantum Mechanics* (McGraw-Hill, New York, 1935), Chap. 5.
- ¹⁴G. W. Kellner, Z. Phys. 44, 91 (1927).
- ¹⁵K. Frankowski and C. L. Pekeris, Phys. Rev. **146**, 46 (1966).
- ¹⁰P. J. Schreiber, R. P. Hurst, and T. E. Duvall, Phys. Rev. A **38**, 3200 (1988).