

High-pressure synthesis of new filled skutterudite compounds SrT4As12 (T = Fe, Ru, Os)

メタデータ	言語: eng
	出版者: Japan Society of Applied Physics
	公開日: 2018-08-23
	キーワード (Ja):
	キーワード (En):
	作成者: 西根, 康平, 川村, 幸裕, 林, 純一, 関根, ちひろ
	メールアドレス:
	所属:
URL	http://hdl.handle.net/10258/00009678

High-pressure synthesis of new filled skutterudite compounds SrT_4As_{12} (T = Fe, Ru, Os)

Kohei Nishine*, Yukihiro Kawamura, Junichi Hayashi, and Chihiro Sekine*

Muroran Institute of Technology, Muroran, Hokkaido 050-8585, Japan

*E-mail: 15043051@mmm.muroran-it.ac.jp; sekine@mmm.muroran-it.ac.jp

We have succeeded in synthesizing samples of new filled skutterudite compounds SrT_4As_{12} (T=Fe, Ru, and Os) using a high-pressure synthesis technique. These compounds have lattice constants of 8.351, 8.521, and 8.561 Å, respectively. The physical properties of the compounds are reported for the first time. The temperature dependences of the electrical resistivity for SrT_4As_{12} (T=Fe, Ru, and Os) exhibit metallic behaviors. The electrical resistivity and magnetic measurements indicates that $SrOs_4As_{12}$ is a new superconductor with a transition temperature of 4.8 K. A broad maximum of magnetic susceptibility at around 50 K and a large electronic specific heat coefficient of 58 mJ mol⁻¹ K⁻² suggest that $SrFe_4As_{12}$ is a nearly ferromagnetic metal with spin fluctuations of Fe 3d electrons.

1. Introduction

The filled skutterudite compounds AT_4X_{12} (A = alkali metal, alkaline earth metal, lanthanide (Ln), and actinide; T = Fe, Ru, Os, and Pt; X = P, As, Sb, and Ge) crystallize in a body-centered cubic structure with the space group $Im\overline{3}$ (T_h^5 , No. 204) 1, 2). Filled skutterudite compounds, especially compounds including Ln, have attracted considerable attention owing to their wide variety of strongly correlated electron behaviors ^{3, 4)}, such as anomalous metal-insulator transition ^{5, 6)}, multipole ordering ⁷⁻¹¹⁾, and unconventional superconductivity ^{12, 13)}. The various attractive features in these systems are believed to be mainly due to the large hybridization between the f electrons of Ln and the p electrons of X. The d electrons of Fe, Ru, and Os atoms should also play important roles in the physical properties. However, the roles have not been well understood yet. Matsuoka et al. reported that filled skutterudite compounds including the alkaline earth metal AFe_4Sb_{12} (A = Ca, Sr,and Ba) reveal the strong ferromagnetic spin fluctuations of Fe 3d electrons and the properties of nearly ferromagnetic metals ¹⁴⁾. Furthermore, AOs₄Sb₁₂ (A=Sr and Ba) shows the characteristics of enhanced Pauli paramagnets where Os 5d electrons play important roles ¹⁵⁾. The filled skutterudite compounds including alkaline earth metal could be a favorable system for systematically investigating the effect of d electrons in skutterudite systems because alkaline earth ions carry no magnetic moments. However, only preliminary studies have been conducted on the properties of P- and As-based compounds including alkaline earth metal because the compounds are difficult to prepare. A high-pressure synthesis method is adequate for preparing As-based skutterudite compounds ^{16, 17)}. Actually, new As-based alkaline-earth-filled skutterudite compounds such as BaFe₄As₁₂ and BaRu₄As₁₂ have been synthesized using a multianvil high-pressure apparatus ^{18, 19)}. Recently, the magnetic properties of BaFe₄As₁₂ prepared under high pressures have been reported ²⁰⁾. BaFe₄As₁₂ shows the feature of a nearly ferromagnetic metal. In this study, we focused on As-based filled skutterudite compounds including Sr as part of our search for new materials using a high-pressure synthesis technique.

2. Experimental methods

The polycrystalline compounds SrT_4As_{12} (T=Fe, Ru, and Os) were prepared at high

temperatures and high pressures using a Kawai-type double-stage multianvil high-pressure apparatus $^{21)}$. We used eight tungsten carbide cubes with a truncated edge length of 11 mm as second-stage anvils. The sample container, made of magnesia (MgO + 5% Cr₂O₃), was transformed into an octahedron with an edge length of 18 mm. The starting materials were placed in a crucible made of boron nitride (BN). The crucible with a graphite heater surrounded with a zirconia (ZrO₂) thermal insulator was inserted into the magnesia octahedron. The samples were prepared by reacting stoichiometric amounts of 3N (99.9% pure)-Sr chip, 4N-Fe, 4N-Ru, 4N-Os, and 6N-As powders at 4 GPa. The reaction temperature was 830-900 °C. The prepared samples were characterized by powder X-ray diffraction using Co $K\alpha_1$ radiation and silicon as a standard. Resistivity was measured by a standard dc four-probe method. Magnetization and dc magnetic susceptibility were measured by a superconducting quantum interference device magnetometer (Quantum Design MPMS). Specific heat measurement was carried out by a thermal relaxation method (Quantum Design PPMS).

3. Results and discussion

Figure 1 shows the powder X-ray diffraction patterns of the new filled skutterudite compounds SrT_4As_{12} (T=Fe, Ru, and Os) synthesized at 4 GPa. Although a small amount of impurity phase [TAs_2 (T = Fe, Ru, and Os)] was detected, most of the observed diffraction peaks were indexable using the skutterudite structure. The lattice constants of the compounds determined by a least-squares fit to the data were 8.351, 8.521, and 8.561 Å, respectively. Figure 2 shows the lattice constants of the new filled skutterudite compounds SrT_4As_{12} (T = Fe, Ru, and Os) along with the values reported earlier for BaT_4As_{12} (T = Fe and Ru) and the rare-earth filled skutterudite arsenides LnT_4As_{12} $^{18, 19, 22-24)}$. Starting with La, there is a typical reduction in lattice constant owing to the contraction of the trivalent ionic radii of the lanthanides with increasing atomic number. The small lattice constants of Ce compounds are due to a strong c-f hybridization. The large lattice constants of the Eu compounds suggest that Eu ions are in a divalent or mixed-valence state. The large lattice constants of BaT_4As_{12} (T = Fe and Ru) and SrT_4As_{12} (T = Fe, Ru, and Os) are considered to be due to the divalent ions of alkaline earth metals.

Figure 3 shows the temperature dependences of electrical resistivity $\rho(T)$ for SrT_4As_{12}

(T = Fe, Ru, and Os). $\rho(T)$ exhibits a positive temperature dependence, such as that observed in, typical metals below room temperature. Figure 4 shows the temperature dependence of the relative electrical resistivity $[\rho(T)-\rho_0]/\rho_{290 \text{ K}}$ for SrT_4As_{12} (T=Fe, Ru, and Os). ρ_0 and $\rho_{290 \text{ K}}$ are the residual resistivity (1.85, 1.12, and 0.21 m Ω cm) and the resistivity at T = 290 K, respectively. SrRu₄As₁₂ shows no anomaly down to 2 K. On the other hand, a broad shoulder in $\rho(T)$ was observed for SrFe₄As₁₂ and SrOs₄As₁₂. In contrast with the monotonic decrease in $\rho(T)$ for SrRu₄As₁₂, a strong curvature in $\rho(T)$ was observed for SrFe₄As₁₂ and SrOs₄As₁₂. The behavior of SrFe₄As₁₂ seems to be a common feature among the alkaline-earth filled skutterudite compounds including Fe, such as $BaFe_4As_{12}$ and AFe_4Sb_{12} (A = Ca, Sr, and Ba) ^{14, 20)}. The curvature in $\rho(T)$ for SrFe₄As₁₂ is stronger than that for an isostructural compound BaFe₄As₁₂ ²⁰⁾. This suggests that the effect of 3d electron spin fluctuation in SrFe₄As₁₂ could be larger than that in BaFe₄As₁₂. On the other hand, the strong curvature of SrOs₄As₁₂ may be due to a strong electron–phonon interaction, which was discussed for AOs_4Sb_{12} (A = Ca, Sr, and Ba) ¹⁵⁾. Matsuoka *et al.* proposed that the strong electron–phonon interaction in the presence of tunneling states of A^{2+} ions in the $[Os_4Sb_{12}]^{2-}$ cage is the origin of a similar shoulder observed in the resistivity of AOs_4Sb_{12} (A = Ca, Sr, and Ba) ¹⁵⁾. Figure 5 shows the $\rho(T)$ of SrOs₄As₁₂ at low temperatures. A sharp drop of $\rho(T)$ was observed below 4.8 K. Figure 6 shows the temperature dependence of the dc susceptibility measured at B = 0.001 T for SrOs₄As₁₂. The susceptibility of the arsenide sharply decreases at around 4.8 K. The hysteresis between zero-field cooling (ZFC) and field cooling (FC) indicates that the arsenide is a type-II superconductor.

Figures 7 and 8 show the magnetic properties of SrFe₄As₁₂. The magnetic behavior of SrFe₄As₁₂ is similar to that of BaFe₄As₁₂ ²⁰⁾. The magnetic parameters of SrFe₄As₁₂ are summarized in Table I with those of BaFe₄As₁₂ and Sb-based compounds as references. The temperature dependences of the magnetic susceptibility $\chi(T)$ and inverse magnetic susceptibility $\chi^{-1}(T)$, measured at B = 1 T, are depicted in Fig. 7. At high temperatures above 150 K, the linear slope of $\chi^{-1}(T)$ yields an effective magnetic moment of 1.36 μ_B /Fe and a Weiss temperature θ_p of 36 K. In contrast with the large negative θ_p (-57 K) for BaFe₄As₁₂²⁰⁾, SrFe₄As₁₂ has a positive θ_p , while the effective magnetic moments of both compounds are almost the same. The magnetization data of SrFe₄As₁₂ up to 7 T at 2 K are depicted in Fig. 8. The magnetization increases linearly with magnetic field. At 7 T, the observed magnetic

moment is $0.045~\mu_B/Fe$. This value is approximately twice as large as that of BaFe₄As₁₂. The small observed magnetic moment at a high magnetic field compared with the effective magnetic moment is also a common feature among the alkaline-earth filled skutterudite compounds including Fe. This indicates that the system is considered as an itinerant magnetic metal. A broad maximum is also observed for SrFe₄As₁₂ at around 50K as shown in Fig. 7. A similar behavior is observed for BaFe₄As₁₂ and AFe₄Sb₁₂ (A = Ca, Sr, and Ba) ^{14, 20)}. The peak of SrFe₄As₁₂ is prominent compared with that of BaFe₄As₁₂. This tendency is the same as that observed in Sb-based compounds ¹⁴⁾. Such a broad maximum was found in nearly ferromagnetic compounds such as Co-based Laves phase compounds, which exhibit a metamagnetic transition ²⁵⁾. Therefore, high-field magnetization measurements are required in clarifying this point.

The temperature dependence of specific heat C(T) for $SrFe_4As_{12}$ at zero field is shown in Fig. 9. The C(T) data exhibit no anomaly due to any phase transitions down to 2 K. Figure 10 shows a plot of C(T)/T versus T^2 at low temperatures. The C(T)/T data of $SrFe_4As_{12}$ can be expressed as $C(T)/T = \gamma + \beta T^2$ with the electronic specific heat coefficient $\gamma = 58$ mJ mol⁻¹ K⁻² and Debye temperature $\Theta_D = (12\pi 4nR/5\beta)^{1/3} = 360 \text{ K}$, where R is the gas constant and n = 17. For many filled skutterudite compounds, the lattice specific heat is modeled by a combination of the usual Debye mode and an Einstein mode due to rattling ²⁶⁾. Figure 11 shows the temperature dependence of $(C - \gamma T)/T^3$ for SrFe₄As₁₂ along with the data of BaFe₄As₁₂ (taken from Ref. 20). We can easily see the contribution of the Einstein mode in this plot. The Einstein specific heat leads to a broad maximum in $(C - \gamma T)/T^3$ at $\Theta_E/4.92$, where Θ_E is the Einstein temperature. We estimated $\Theta_E = 133$ K for SrFe₄As₁₂. For BaFe₄As₁₂, a broad maximum related to the Einstein mode was not observed ²⁰⁾. The results could be closely related to the difference between the ionic radii of Ba and Sr. Namely, a smaller Sr ion shows a random motion in the large space inside the cage. On the other hand, there is no space to move inside the cage in BaFe₄As₁₂. The thermal parameters of SrFe₄As₁₂ are also summarized in Table I together with those of BaFe₄As₁₂ and Sb-based compounds as reference compounds.

4. Conclusions

Using a high-pressure synthesis technique, we have succeeded in synthesizing the new

skutterudite compounds SrT_4As_{12} (T=Fe, Ru, and Os) with lattice constants of 8.351, 8.521, and 8.561 Å, respectively. $SrOs_4As_{12}$ was found to be a new superconductor at around 4.8 K. The magnetic susceptibility of $SrFe_4As_{12}$ shows a broad peak at around 50 K and a large electronic specific heat coefficient is observed for the compound. The results suggest that $SrFe_4As_{12}$ is a nearly ferromagnetic metal with spin fluctuations of Fe 3d electrons similar to $BaFe_4As_{12}$. The new synthesized As-based filled skutterudite compounds including Sr could be favorable systems for systematically investigating the effect of d electrons in skutterudite systems. Furthermore, $EuFe_4As_{12}$ with divalent Eu^{2+} ions was reported to exhibit a ferrimagnetic-like phase transition with an extremely high transition temperature of $152 K^{27}$). The exchange interaction between the 3d electrons of Fe and the 4f electrons could be essential to the ordering in this compound. Therefore, $SrFe_4As_{12}$ and $BaFe_4As_{12}$ with divalent ions, without 4f electrons, are important reference compounds for elucidating the origin of the magnetic properties of $EuFe_4As_{12}$ in detail.

Acknowledgment

This work was supported by JSPS KAKENHI Grant Number 23340092.

References

- 1) W. Jeitschko and D. Braun, Acta Crystallogr., Sect. B 33, 3401 (1977).
- E. Bauer, A. Grytisiv, Xing-Qiu Chen, N. Melnychenko-Koblyuk, G. Hilscher, H. Kaldarar, H. Michor, E. Royanian, G. Giester, M. Rotter, R. Podloucky, and P. Rogl, Phys. Rev. Lett. 99, 217001 (2007).
- 3) See Proc. Int. Conf. New Quantum Phenomena in Skutterudite and Related Systems, J. Phys. Soc. Jpn. 77, Suppl. A (2008).
- 4) H. Sato, H. Sugawara, Y. Aoki, and H. Harima, in *Handbook of Magnetic Materials*, ed. K. H. J. Buschow (North-Holland, Amsterdam, 2009) Vol. 18, Chap. 1.
- 5) C. Sekine, T. Uchiumi, I. Shirotani, and T. Yagi, Phys. Rev. Lett. 79, 3218 (1997).
- 6) C. Sekine, T. Uchiumi, I. Shirotani, and T. Yagi, in *Science and Technology of High Pressure*, ed. M. H. Manghnani, W. J. Nellis, and M. F. Nicol (Universities Press, Hyderabad, 2000) p. 826.
- 7) M. Yoshizawa, Y. Nakanishi, M. Oikawa, C. Sekine, I. Shirotani, S. R. Saha, H. Sugawara, and H. Sato, J. Phys. Soc. Jpn. **74**, 2141 (2005).
- 8) M. Yoshizawa, Y. Nakanishi, T. Kumagai, M. Oikawa, C. Sekine, and I. Shirotani, J. Phys. Soc. Jpn. **73**, 315 (2004).
- 9) Y. Kuramoto, Prog. Theor. Phys. Suppl. **176**, 77 (2008).
- 10) A. Kiss and Y. Kuramoto, J. Phys. Soc. Jpn. **78**, 124702 (2009).
- 11) R. Shiina, J. Phys. Soc. Jpn. **82**, 083713 (2013).
- 12) E. D. Bauer, N. A. Frederick, P. -C. Ho, V. S. Zapf, and M. B. Maple, Phys. Rev. B **65**, 100506 (2002).
- 13) Y. Aoki, T. Tayama, T. Sakakibara, K. Kuwahara, K. Iwasa, M. Kohgi, W. Higemoto, D.E. MacLaughlin, H. Sugawara, and H. Sato, J. Phys. Soc. Jpn. **76**, 051006 (2007).
- 14) E. Matsuoka, K. Hayashi, A. Ikeda, K. Yanaka, T. Takabatake, and M. Matsumura, J. Phys. Soc. Jpn. **74**, 1382 (2005).
- E. Matsuoka, S. Narazu, K. Hayashi, K. Umeo, and T. Takabatake, J. Phys. Soc. Jpn. 75, 014602 (2006).
- T. Namiki, Y. Aoki, H. Sato, C. Sekine, I. Shirotani, T. D. Matsuda, Y. Haga, and T. Yagi, J. Phys. Soc. Jpn 76, 093704 (2007).
- 17) C. Sekine, R. Abe, K. Takeda, K. Matsuhira, and M. Wakeshima, Physica B 403, 856

(2008).

- 18) K. Takeda, K. Ito, J. Hyashi, C. Sekine, and T. Yagi, J. Phys. Soc. Jpn. 80, SA029 (2011).
- K. Takeda, N. Hoshi, J. Hayashi, C. Sekine, S. Kagami, I. Shirotani, and T. Yagi, J. Phys.: Conf. Ser. 215, 012130 (2010).
- 20) C. Sekine, T. Ishizaka, K. Nishine, Y. Kawamura, J. Hayashi, K. Takeda, H. Gotou and Z. Hiroi, Phys. Procedia **75**, 383 (2015).
- 21) C. Sekine, T. Kawata, Y. Kawamura, and T. Yagi, J. Korean Phys. Soc. 63, 359 (2013).
- 22) D. J. Braun and W. Jeitchko, J. Solid State Chem. 32, 357 (1980).
- 23) W. Jeitchko, A. J. Foecker, D. Paschke, M. V. Dewalsky, C. B. H. Evers, B. Kunnen, A. Lang, G. Kotzyba, U. C. Rodewald, and M. H. Moller, Z. Anorg. Allg. Chem. **626**, 1112 (2000).
- 24) A. Rudenko, Z. Henkie, and T. Cichorek, Solid State Commun. 242, 21 (2016).
- 25) H. Yamada, Phys. Rev. B 47, 11211 (1993).
- 26) K. Matsuhira, C. Sekine, M. Wakeshima, Y. Hinatsu, T. Namiki, K. Takeda, I. Shirotani, H. Sugawara, D. Kikuchi, and H. Sato, J. Phys. Soc. Jpn. **78**, 124601 (2009).
- 27) C. Sekine, K. Akahira, K. Ito, and T. Yagi, J. Phys. Soc. Jpn. 78, 093707 (2009).

Figure Captions

Fig. 1. (Color online) Powder x-ray diffraction patterns of SrT_4As_{12} (T = Fe, Ru, and Os) prepared under high pressures.

Fig. 2. (Color online) Lattice constants of SrT_4As_{12} (T = Fe, Ru, and Os) and BaT_4As_{12} (T = Fe and Ru) and rare-earth filled skutterudite arsenides LnT_4As_{12} (Ln = Fe rare-earth elements; T = Fe, Ru, and Os).

Fig. 3. (Color online) Temperature dependence of electrical resistivity $\rho(T)$ of SrT_4As_{12} (T = Fe, Ru, and Os).

Fig. 4. (Color online) Temperature-dependent part $\rho(T)$ - ρ_0 of electrical resistivity for SrT_4As_{12} (T = Fe, Ru, and Os), normalized at 290K.

Fig. 5. (Color online) Temperature dependence of electrical resistivity $\rho(T)$ of SrOs₄As₁₂ at low temperatures.

Fig. 6. (Color online) ZFC and FC dc magnetic susceptibility χ at 0.001 T for SrOs₄As₁₂.

Fig. 7. (Color online) Magnetic susceptibility χ and inverse magnetic susceptibility χ^{-1} versus temperature for SrFe₄As₁₂ at 1 T.

Fig. 8. (Color online) Magnetization for SrFe₄As₁₂ at 2 K.

Fig. 9. (Color online) Temperature dependence of the specific heat C for $SrFe_4As_{12}$.

Fig. 10. (Color online) T^2 dependence of C/T for SrFe₄As₁₂.

Fig. 11. (Color online) Temperature dependence of the specific heat for $SrFe_4As_{12}$ and $BaFe_4As_{12}$. The data are displayed in the form of $(C-\gamma T)/T^3$. The data of $BaFe_4As_{12}$ were taken from Ref. 20.

Fig.1. (Color Online)

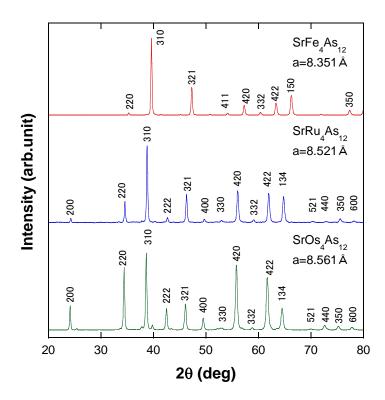


Fig.2. (Color Online)

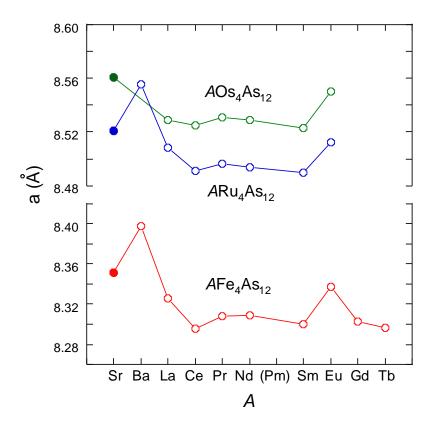


Fig.3. (Color Online)

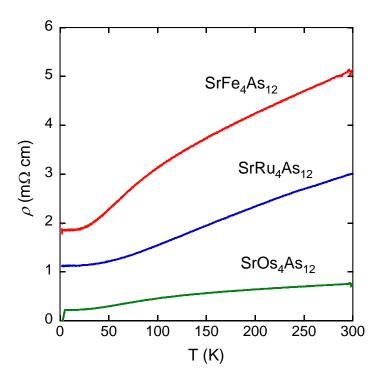


Fig.4. (Color Online)

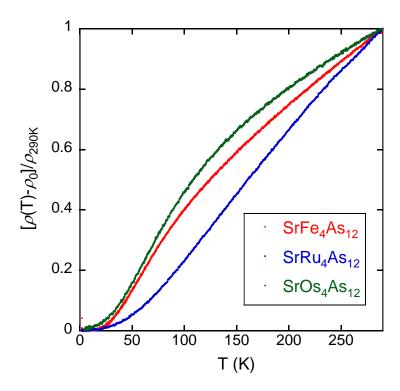


Fig.5.

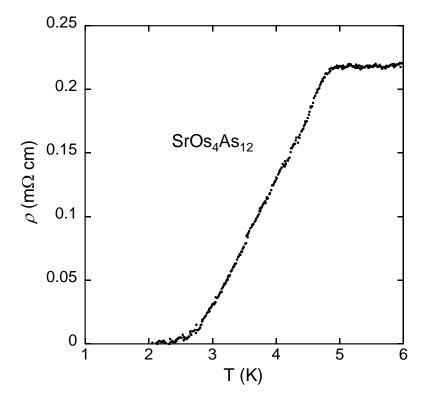


Fig.6. (Color Online)

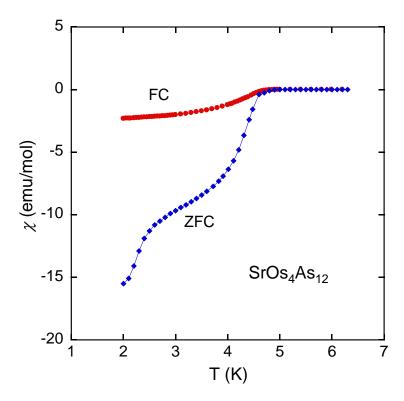


Fig.7. (Color Online)

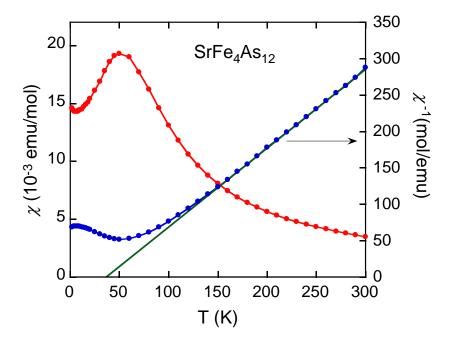


Fig.8.

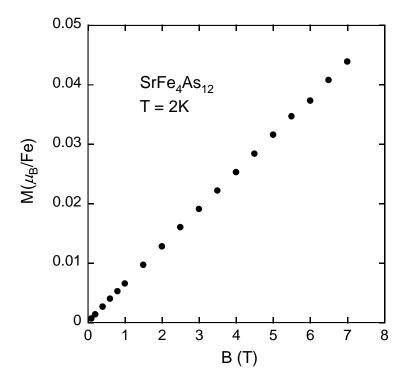


Fig.9.

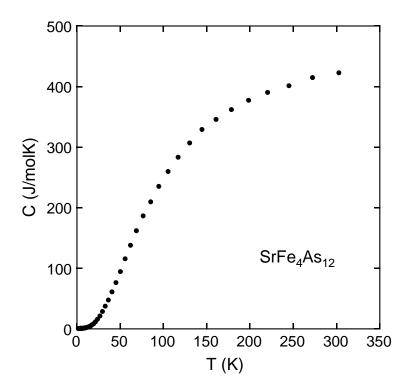


Fig.10.

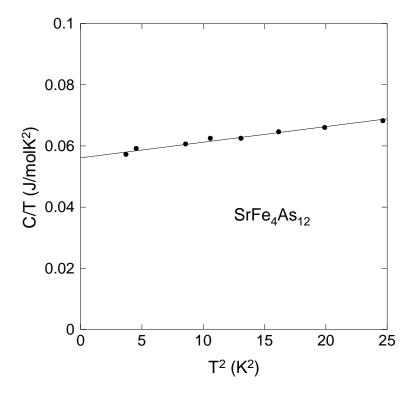


Fig.11. (Color Online)

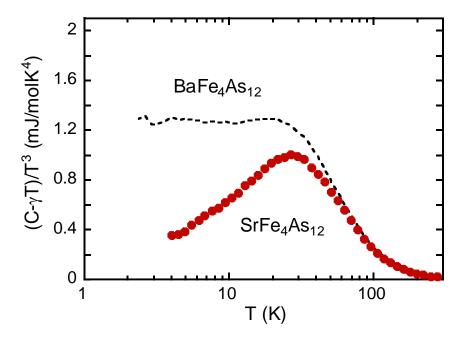


Table I. Magnetic and thermal parameters of SrFe₄As₁₂, BaFe₄As₁₂, and Sb-based compounds as reference compounds (μ_{eff} : effective magnetic moment, θ_P : Weiss temperature, γ : electronic specific heat coefficient, Θ_D : Debye temperature, and Θ_E : Einstein temperature).

_	Compound	$\mu_{ m eff}$	$\theta_{ ext{P}}$	γ	$\boldsymbol{arTheta}_{ m D}$	$\boldsymbol{arTheta}_{ ext{E}}$	Reference	
		$(\mu_{ m B})$	(K)	$(mJ/molK^2)$	(K)	(K)		
_	SrFe ₄ As ₁₂	1.36	36	58	360	133	This work	
	$BaFe_{4}As_{12} \\$	1.46	-57	62	295	-	20	
	$SrFe_4Sb_{12} \\$	1.47	53	87	279	103	14, 15	
	$BaFe_{4}Sb_{12} \\$	1.50	31	104	298	108	14, 15	
	SrFe ₄ Sb ₁₂	1.47	53	87	279	103	14, 15	